## Bristol ChemSpeed Robotic Synthesis – Experiment/Project Application Form

Please fill in the details below and submit this form to: **XXXXXXXXXX**.

After validating your proposal we will contact you to discuss scheduling, costs and practicalities of your experiments. Please note the following:

- Users must provide all chemicals and consumables required to conduct experiments
- Academic users will be trained to run their own experiments by UoB technical support
- All reaction data will be stored and retained following UoB standard protocols
- Users will be requested to contribute to cost of repairs arising from their usage.

Name: James Fordham and Rory Mykura

**Group/Company:** University of Bristol

**Email Address:** example@bristol.ac.uk

**Experiment/Project Name:** Automated Click Chemistry – Synthesis of a Triazole Library

Grant code/cost centre: XXXXXXX

(Users must pay consumables costs for their reactions. Academic projects will not be charged for usage time but future **grant proposals must include costings** for this system).

Aims of the experiments (proof-of-principle, catalyst/conditions screen, etc) and any pertinent details e.g. when would you like to conduct the experiments, what are the likely benefits of the robot etc.

Library Synthesis: Synthesis of a Triazole Library

Screening of different starting boronic esters and alkynes in a fully optimized fashion

Desired date of use: 14/1/19 to 18/1/19.

## Synthetic scheme, including reagents and conditions/times (add additional sheet if required) and anticipated workflow

Synthetic scheme:

- Number of variables (e.g. reagent, catalyst, stoichiometry) to be explored?
  2 variables = 8 boronic acids and 4 alkynes
- Total number of reactions?

8 boronic acids x 4 alkynes = 32 reactions

• Scale of reactions? (Check description of equipment)

0.05 mmol of Reagent B, total volume = 0.15 mL

• Reaction time?

10 h

• Solid dispensing? (Check description of equipment)

Yes,  $32 \times (5.0 - 10.0 \text{ mg})$  for the boronic acid  $32 \times (10.0 \text{ mg})$  for the sodium azide

Please conduct a full risk assessment (attached separately) and note here any particular safety/risk concerns you have identified

Use of sodium azide